## Time-schedule of lectures on biochemistry for «English medium» students Fall semester 2022-2023

Lectures are given on TUE (lecture hall No2408)
\* Lecture is given im Moodle

Head of department of clinical biochemistry, microbiology and laboratory diagnostics, professor

T.A. Bazhukova

## Time-table of practical classes on biochemistry for students of ENGLISH MEDIUM program. Biomedical chemistry department. Fall semester 2022/2023 year.

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	03-07.10	Proteins as biopolymers	-	Proteins: definition of class.
			2)	Biological functions of proteins.
			3)	Classification of proteins.
			4)	Aminoacid composition of proteins.
	\		5)	Physico-chemical and chemical properties of aminoacids.
2 1	10-14.10	Proteins structure and physico-chemical		Polypeptide theory of protein structure.
		properties.	2.	Conformation and configuration of polypeptide chain.
	1 1 1	is at sing and way are and a plant of the	<u>ن</u>	Levels of structural organisation of proteins: primary, secondary, tertiary and quaternary structure
	4. 2.	I Bellemona seas	4.	Native conformation and functional sites of protein molecules.
ž.	- <del>-</del>	amoleculo, pro programa de la complexes	5.	Supramolecular protein complexes.
	· ballings	ico-chemical properties of processy		Physico-chemical properties of proteins.
rises of 1	17-21.10	Chemical and biologycal properties of	pv.I	Chemical properties of proteins (colour reactions, hydrolysis, phosphorylation, glycosylation, gamma-
niced tions	carte	proteins. Protein-ligand interactions.	300	carboxylation, methylation), biological significance.
	24 Prox	2.57	2. n	2. mProtein-ligand interaction as a basis for biological functions of proteins.
	3 LAB:	_	3. L	3. LAB: Determination of protein concentration in serum by a) refractometry and 2) biuret method.
4 2	24.10- VIII	"Vitamins. The biological role of	1	Vitamins: definition and classification.
2	28.10		2.	Fat-soluble vitamins, functions.
	35- D)	er-soluble vitamos l'unctions.	ω.	Water-soluble vitamins, functions.
	dir.	in of action		Antivitamins, mechanism of action.
	121	mi disbalanci oviiamino aviismin		Vitamin disbalance: hypovitaminosis, avitaminosis and hypervitaminosis.
	(AB	clambation if Vit Community in fi		LAB: Determination of VitiC concentration in food (fruits and vegetables) and urine.
5	31-	Enzymes: structure, properties, biological		Enzymes: definition and biological role.
0	03.11,	role.	4	Common features and difference of enzymes and non-biological catalysts.
1	11.11*		w	Chemical nature of enzymes. Simple and complex enzymes.
			4.	Cofactors: classification and role in the catalysis.
		200 Comment of the Co		Functional sites of enzymes (active, regulatory, contact platforms).
	*		6.	Main properties of enzymes: high catalytical activity, specificity of action, dependence of their activity on
				enzyme and substrate concentration, temperature, pH.
			7.	Lab: properties of enzymes.
6 0	07-	Enzymes: mechanism of action,	1.	Mechanism of enzyme action (steps of catalysis, thermodynamic parameters, substrate modification in the active
	10.11,	regulation of activity. Medical		site, etc.).
1	18.11*	enzymology. Final test	2.	Regulation of enzymatic activity: purpose. Fast and slow mechanisms.
				Main variants of fast regulatory mechanisms: allosteric modulation, covalent modification, protein-protein
			4	meraction, activation of proemzymes, competitive inhibition.  Medical enzyme/force enzyme nathology enzyme diagnostics, enzyme therapy
				intention only into barriously, only into diagnostics, only into a large metapy.

	7	14_17 11	Enarmy matchalism Patahalism	-	Town of thomas demands
		25.11*	Biological oxidation.	2:	Exergonic and endergonic reactions.
					Macroergic compounds.
				4.	Catabolism of nutrients.
				9 %	Biological oxidation: types of oxidative reactions, enzymes, functions.  Lab: determination of ATP in muscles
	8	21-24.11,	Mitochondrial oxidation.	-	Krebs cycle: general description, reactions, regulation, biological role, energy balance.
		02.12*		2.	Electron transporting chain: structural organisation, description of its work.
				į.	Redox potential difference as driving force for electron transfer. Energy balance for oxidation of NADH or
			N		FADH <sub>2</sub> .
			The Monte The American	4.	Oxidative phosphorylation as mechanism of ATP production.
			THE STATE OF THE S	5.	ETC inhibitors.
1		26.	APPER MC Tanish	6.	Uncoupling mechanism.
GT:	9	28.11-			Hypoenergetical states. Their origin.
55	Sac.	01112,	200	2.	Microsomal oxidation. Mono- and dioxygenase reactions. Cytochrome P450. Functions of microsomal oxidation.
		09.12*	(C)	3	Reactive oxygen species (ROS). Mechanism of synthesis, their action in normal and pathological conditions,
100		MZC INSC	ivation (encomatic and non encomatic ways	1,673	inactivation (enzymatic and non- enzymatic ways).
mo	10	05-08.12, Car			Carbohydrates: definition, classification, functions.
HISC	negen	16.12* De	digestion and absorbtion. Glycogens and	2.	Digestion of food carbohydrates: enzymes, products.
CH	1		ids in the intestinal'v	3	Absorbtion of monosaccharids in the intestinal wall.
146	27 88	# + F0			Pool of glucose in the body.
		(-1). (-1).	and break fown		Glycogen metabolism (synthesis and breakdown) in the liver, its regulation by hormons,
	100		Determination of anylose offerty in salva	6.	Lab: Determination of amylase activity in saliva.
3Cc	112	12-15.12,	Metabolic pathways of intracellular	1.0	1. Dichotomic oxidation of glucose in aerobic and anaerobic conditions, energy balance, biological significance of
	71.	23.12*	carbohydrates exchange.		these processes.
		1.9	Thorofe Wese and to use.	2.	Oxidation of galactose and fructose.
				3	Conversion of glucose to derivatives of glucuronic acid.
				4.	Pentosephosphate pathway of glucose oxidation: reactions of oxidative step, biological significance of the
			(A		process.
		- 1		5.	Gluconeogenesis: reactions, function, regulation.
				LA	LAB: Determination of glucose concentration in blood by glucose oxidase test.
***	12	19-22.12,	Regulation of carbohydrate exchange.		Hormons with hypo- or hyperglycemic effect (insulin, glucagon, adrenalin, cortisol, STH, thyroxin). Mechanisms
		30.12*	Diseases. Final test		of their influence on glucose concentration in blood.
				2.	Primary (lactase deficiency, galactosemia [African and Swiss variants], fructose intolerance and essential
					fructosuria, glycogen exchange diseases) and secondary (diabetes mellitus) diseases caused by (or resulted from)
					abnormal metabolism of carbohydrates.
					Methods of carbohydrates exchange investigation.
				4.	MCQ test on topic "Energy metabolism. Carbohydrates: structure, functions and metabolism"

\* for groups on Friday
Зав.кафедрой клинической биохимии, микробиологии и лабораторной диагностики д.м.н., проф. Утверждено на заседании кафедры 30.08.2022 г., протокол №1

Дамуя (Т.А. Бажукова)